Lewis Acid-mediated Highly Regioselective Ring-expansion of Methyl 2-Phenyl-1-(arylhydroxymethyl)cyclopropanecarboxylates

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A novel ring-expansion of methyl (arylhydroxymethyl)cyclopropanecarboxylates 1 using Sc(OTf)₃ or BF₃·OEt₂ afforded 1,2-dihydronaphthalene-3-carboxylic acid ester 2 in high to excellent yields. In the reaction, highly regioselective ring opening of cyclopropane and sequential cyclization occurred.

1-Aryl-1,2-dihydronaphthalene analogs are attracting considerable attention due to their distribution in nature (for examples, trilobatin A, B, cyclogalgravin, and magnoshinin, Scheme 1), multiple biological activities, and usefulness as important synthetic intermediates.¹ As a part of our ongoing program of synthetic studies on the transformation of gemdihalocyclopropanes,² we have recently reported highly stereoselective SmI₂-promoted Reformatsky-type reactions of 1chlorocyclopropanecarboxylate to afford (arylhydroxymethyl)cyclopropanecarboxylates 1 (Scheme 2).³ Here we report a Lewis acid-mediated highly regioselective ring expansion of methyl (arylhydroxymethyl)cyclopropanecarboxylates 1^4 to give 1-aryl-1,2-dihydronaphthalene-3-carboxylic acid esters 2.⁵

Initially, we investigated the reaction of cyclopropanecarboxylate 1a with various Lewis acids. Table 1 lists the results of the ring-expansion. TiCl₄, SnCl₄, or TBDMSOTf promoted ringexpansion to afford dihydronaphthalene 2a in low to moderate yields (Entries 1-4). These results are inconsistent with those for benzannulation of gem-dichlorocyclopropylmethanol.2b Under Seebach's condition,⁵ no amount of ester 2a was obtained (Entry 5). In this case, hydrolysis of ester occurred along with the ring-expansion to give corresponding 1,2-dihydronaphtha-



Next we investigated the ring-expansion of (arylhydroxymethyl)cyclopropanecarboxylates 1b-1e with Sc(OTf)₃ at 83 °C. As expected, the ring-expansion of ester 1b-1e proceeded smoothly to afford dihydronaphthalene 2b-2e in good to high yields (Table 2, Entries 1-4). In the case of 1d, hydrolysis of ester 2d occurred as a side reaction (Entry 3). In addition, we also investigated similar reaction of diaryl analogs 1f-1i (Table 3). Every case of ester 1f-1i underwent the desired ring-expansion to give dihydronaphthalene 2f-2i in good to excellent yields. Treatment of 1f with Sc(OTf)₃ at 83 °C afforded dihydronaphthalene 2f in high yield (Entry 1). The use of Yb(OTf)₃ or TiCl₄ decrease the yield of ring-expansion (Entries 2 and 3). In the case of diaryl analog 1f, BF₃•Et₂O also promoted the ring-expansion in excellent yield (Entry 4).



Lewis Acid

Temp/°C

rt

83

rt

rt

35

rt

rt

83

rt

83

rt

83

^aReactions were carried out under an Ar atmosphere. ^bReaction

time is 1 h. c1.1 equiv of Lewis acid was used. dA mixture of

diastereo isomers (ratio = 1/1) was used for the reactions of **1a**.

^eIsolated. ^fBased on Ref. 5, 1000 equiv of H₂SO₄ was used.

1a

Acid

TiCl₄

TiCl₄

SnCl₄

TBDMSOTf

H₂SO₄^f

CF₃CO₂H

BF3.Et2O

BF3.Et2O

Yb(OTf)₃

Yb(OTf)₃

Sc(OTf)₃

Sc(OTf)₃

Entry

1

2

3

4

5

6

7

8

9

10

11

12



Scheme 1.



Scheme 2.

CO₂Me

Yield^e/%

43

36

25

54

0

10

70

62

10

90

66

95

2a

Solvent

CH₂Cl₂

EDC

CH₂Cl₂

CH₂Cl₂

Et₂O

none

CH₂Cl₂

EDC

CH₂Cl₂

EDC

 CH_2Cl_2

EDC

Table 2. Highly regioselective $Sc(OTf)_3$ -mediated ring-expansion of methyl (arylhydroxymethyl)cyclopropanecarboxylates $1b-1e^{a-d}$



^aReactions were carried out under an Ar atmosphere. ^bReaction time is 1 h. ^c1.1 equiv of Lewis acid was used. ^dA mixture of diastereo isomers (ratio = 1/1) was used for the reactions of **1a**. ^eIsolated.

Table 3. Highly regioselective ring-expansion of methyl (diarylhydroxymethyl)cyclopropanecarboxylates $1f-1i^{a-c}$



Entry	Substrate	\mathbb{R}^1	Acid	Temp /°C	Solvent	Product	Yield [®] /%
1	1f	Н	Sc(OTf) ₃	83	EDC	2f	92
2	1f	Η	Yb(OTf) ₃	83	EDC	2f	85
3	1f	Η	TiCl ₄	rt	CH_2Cl_2	2f	60
4	1f	Η	$BF_3 \cdot Et_2O$	rt	CH_2Cl_2	2f	99
5	1g	Cl	$BF_3 \cdot Et_2O$	rt	CH_2Cl_2	2g	68
6	1h	Me	$BF_3 \cdot Et_2O$	rt	CH_2Cl_2	2h	82
7	1i	OMe	$BF_3 \cdot Et_2O$	rt	CH_2Cl_2	2i	71

^aReactions were carried out under an Ar atmosphere. ^bReaction time is 1 h. ^c1.1 equiv of Lewis acid was used. ^dIsolated.

Similar reaction of diaryl analogs **1g–1i** proceeded in good to high yields (Entries 5–7).

The proposed mechanism of the ring-expansion mediated by $Sc(OTf)_3$ is as follows (Scheme 3). First, $Sc(OTf)_3$ chelates with the OH and carbonyl of β -hyrdoxyester 1 to give an intermediate 3. Successive $Sc(OTf)_3$ -promoted elimination of the OH group gives the cationic intermediate 4. Then, highly regioselective ring-opening and Friedel–Crafts-type cyclization sequentially occur to give dihydronaphthalene 2. Cyclization proceeds smoothly in the Z-intermediate 5 of the cation. In the reaction using TiCl₄ or SnCl₄, the presence of Cl⁻ adversely affects the cyclization, causing the chlorination of benzyl cation to occur. BF₃·OEt₂ also promotes the ring-expansion of symmetrically substituted diaryl analog 1f in excellent yield. In this case (R³ = p-R¹-Ph, R² = H, intermediate 5 = 6), the ring-expansion proceeds smoothly without the problem of the E-intermediate.

In conclusion, we developed a novel synthesis of 1-aryl-1,2dihydronaphthalene-3-carboxylic acid esters utilizing a Lewis acid-mediated regioselective ring-expansion of methyl (arylhydroxymethyl)cyclopropanecarboxylate. Application of the



Scheme 3.

present method to a total synthesis of natural product is now being performed.

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- 4 (Arylhydroxymethyl)cyclopropanecarboxylates **1a–1i** were prepared by the similar method of our previous report: see Ref. 3a.
- 5 As a similar reaction, an example has been reported to synthesize 3-bromo-1,1-dimethyl-4-phenyl-2-hydronaphthalene: R. Dammann, D. Seebach, *Chem. Ber.* 1979, *112*, 2167.
- 6 Procedures were described in Supporting Information, which is available electronically on the CSJ-Journal Web site, http://www.csj.jp/journals/chem-lett/index.html.
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